

The growing problem of water pollution with pharmaceuticals. Challenges, prospects and solutions.

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Pharmaceuticals in modern medicine





Fate of pharmaceuticals in the environment



Clean Technologies and Environmental Policy (2020) 22:11-42



Environmental fate of pharmaceuticals





Active pharmaceutical ingredients (API)

Pharmaceutical compound	Molecule	Influent (µg/L)	Effluent (μg/L)	Removal efficiency (%)
Analgesics and anti-inflammatory	Naproxen	5.08	0.93	82
	Ibuprofen	13.48	3.48	74
	Acetaminophen	36.7	0.04	95
Antibiotics	Doxycycline	0.65	0.42	35
	Ciprofloxacin	0.62	0.23	62
	Ofloxacin	0.48	0.17	64
	Sulfamethoxazole	0.32	0.26	18
	Tetracyclin	46.8	2.34	95
Antiepileptics	Carbamazepine	0.73	0.77	not removed
	4-Aminoantipyrine	1.51	0.67	55
	Diclofenac	1.04	0.68	34
Desinfectant	Triclosan	0.85	0.19	77



Carbamazepine

300

200

100

Austria



Finland France Cermany Repetands Swittenland Canada USA Japan Korea South Korea

• Carbamazepine - dibenzazepine derivative with antiepileptic and psychotropic activity, also well established in treatment of severe pain syndromes associated with neurological disorders, such as trigeminal neuralgia.



 Carbamazepine shows low sorption properties and high persistence to biodegradation





Acetaminophen

- Acetaminophen (paracetamol) member of the analgesic and antipyretic drugs group.
- It is included in the WHO Model List of Essential Medicines in 2019. One of the top 200 drugs prescribed overall the world.
- Acetaminophen is used as an antipyretic (fever reducer) and as an analgesic (pain reliever) due to migraine, headache, muscular aches, neuralgia, backache, toothache, and general pain.
- Acetaminophen generate chloramines, which are toxic.
- Acetaminophen gives rise to highly toxic N-acetyl-p-benzoquinone imine and 1,4benzoquinone upon chlorination treatment.
- Detected in wastewater treatment plants, rivers and muds with concentrations in the range from 6 to 65 μ g/L.







Why do we need to improve the quality of surface waters?



Status of the Baltic Sea, related to pharmaceuticals presence (a), pharmaceuticals detected in wastewaters (b), and marine waters (c,d), in the Baltic Sea Region*

*Source: HELCOM, Pharmaceuticals in the aquatic environment of the Baltic Sea region A status report International Initiative on Water Quality-IIWQ, 2017.





Why Advanced Oxidation Processes ?

- biologically toxic or non-degradable materials such as aromatic, pesticides, petroleum etc.
- high concentrated (to increase /BOD/COD/ biodegrability)
- volatile organic compounds in wastewater
- to treat effluent of secondary treated wastewater tertiary treatment





Advanced Oxidation Processes

OH.	

In-situ production of hydroxyl radicals (•OH) and sulfate radicals (•SO₄-)

oxidizing agent	oxidation potential [V]
hydroxyl radicals	2.89
sulfate radicals	2.80
Fenton's reagent	2.76
ozone	2.07
hydrogen peroxide	1.78
hypochlorous acid	1.49
chlorine	1.36

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Recent advances in photocatalysis





The surface chemistry of photocatalytic materials



 $\{1 0 1\}$ 100 nm 500nm

The degradation process over the TiO_2 photocatalyst is mostly controlled by the distribution of surface trapped states, specific for each crystal facet (highest density for the {1 0 1} facets and the lowest for {0 0 1} one).

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Crystal facets engineering of BiVO₄



SEM images of **a**) BVO_D, **b**) SAED patterns of BVO_D, **c**) BVO_M, **d**) BVO_O, **e**) TEM and **f**) HRTEM images of BVO_O with calculated *d*-spacing in inset.



 $\lambda_{exc} = 315 \text{ nm}$

Intensity (arb. units)

350

400

450

500

550

Wavelength (nm)

Photoluminescence spectra

of faceted BVO

{1 2 0} and {0 2 1} facets.

600

650

700

750

cut-off filter 400 nm

-BVO M BVO D

BVOO

Crystal facets engineering of BiVO₄



-BVO O

The order of the kinetics during photocatalysis is BVO_O > BVO_D > BVO_M

70

60

50

(UX) 40 30 Z-

20

10



Peroxymonosulphate-assisted photocatalytic degradation of environmental pollutants





Peroxydisulfate (PDS)

Peroxymonosulfate (PMS)

- PS and PMS assisted processes are efficient in degrading persistent organic pollutants, including pesticides, pharmaceuticals, and plasticizers, attributed to the generation of SO4⁻⁻ and/or ⁻OH
- PMS may effectively produce sulphate radicals with strong oxidation ability.
- The sulphate radical (•SO₄⁻) possesses a close or even higher redox potential of 2.5– 3.1 V (vs. NHE) compared to hydroxyl radical (E₀=1.8-2.8 V). Moreover, sulphate radicals have higher selectivity, longer half-life (30–40 µs), and could be activated in the broader pH range from 2 to 8 than •OH radicals.







Summarizing



- ✓ The surface chemistry and controlled growth of semiconductor nanocrystals
- ✓ The electron-donor cocatalyst loading on the crystal facets with the preference of holes accumulation
- The electron-acceptor cocatalyst for reductive facets
- The hybridization of semiconductor with graphenerelated material and combining photocatalytic reaction with transition metal-based PMS acceleration



Project granted by the European Union within Interreg Baltic Sea Region Programme. "Improving quality of BSR waters by advanced treatment processes", AdvIQwater (2022-2025)

The AdvIQwater project aims to the development of solar-driven photocatalysis, biofilms, and fungal treatment for efficient removal of pharmaceuticals from water.